

TRANSPOSITION OF CO₂ GROUP IN 2-FURANONES CATALYZED BY PALLADIUM(0)

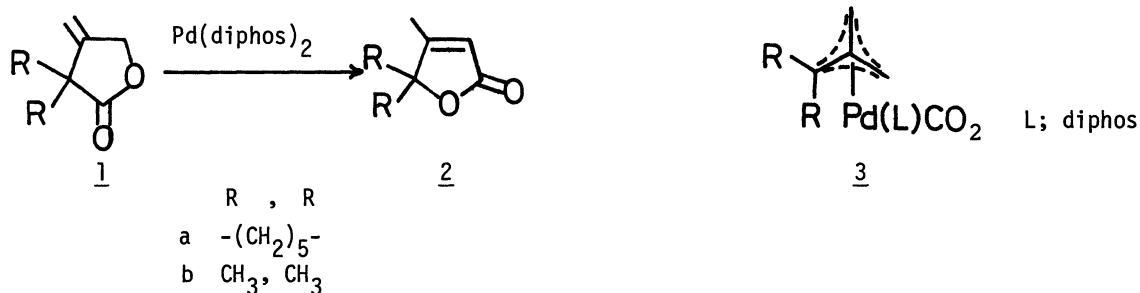
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3,3-Dialkyl-4-methylenedihydro-2(5H)-furanone 1 isomerized into another furanone, 4-methyl-5,5-dialkyl-2(5H)-furanone 2, with transposition of the CO₂ group in the presence of a catalytic amount of Pd(diphos)₂ [diphos=Ph₂PCH₂CH₂PPh₂].

It is known that π-allyl palladium complexes are available from allylic acetate with cleavage of the C-O bond.¹⁾

We have found that rather unique reaction occurs with cleavage of both the C-C and the C-O bonds when a furanone bearing a suitable allylic group is treated with palladium(0): The furanone of the type 1 was converted into another furanone of the type 2 with transposition of CO₂ group in the presence of Pd(diphos)₂.



The furanone of the type 2 are probably derived from the corresponding 4-methylene-furanones by isomerization.

When 3-spiro-1'-cyclohexane-4-methylenedihydro-2(5H)-furanone (1a)²⁾ (0.17 mmol) was treated with an equimolar amount of Pd(diphos)₂ in benzene (10 ml) under CO₂ pressure (60 atm) at 125 °C for 21 h, 4-methyl-5-spiro-1'-cyclohexane-2(5H)-furanone (2a)³⁾ was formed in a good yield (76%; conversion 100%). The same reaction took place with 3,3-dimethyl-4-methylenedihydro-2(5H)-furanone (1b).⁴⁾ And the reaction was found to be catalytic with respect to palladium even under N₂ atmosphere. For example, 1b (1.4 mmol) was treated with Pd(diphos)₂ (0.04 mmol) at 110 °C for 1.5 h in toluene (2 ml) under N₂ atmosphere (1 atm) to yield 4,5,5-trimethyl-2(5H)-furanone (2b)⁵⁾ (0.36 mmol, yield 26%, turnover number 9) with 100% conversion. Under CO₂ pressure (60 atm) the yield of 2b increased to 38%. In these reactions the formation of two other products with molecular weight 218 and 390 was also observed by GLC-mass, the structures of which were uncertain.

$\text{Pd}(\text{PPh}_3)_4$ scarcely catalyzed this reaction.

Although we are not certain the exact pathway of this reaction, the scheme involving a trimethylenemethane complex of the type 3 as an intermediate is a possible one.

References

- 1) For example, B. M. Trost, "Organopalladium Intermediate in Organic Synthesis" (Tetrahedron Report No. 32), Pergamon Press(1978).
- 2) This compound was synthesized⁶⁾ by the reaction of cyclohexylidenecyclopropane and CO_2 in the presence of $\text{Pd}(\text{PPh}_3)_4$.
- 3) IR(neat) 1740(C=O), 1640(C=C), 1210(C-O), and 850(HC=C) cm^{-1} . NMR(CCl_4) δ 1.0-1.9(10 H, br, $-\text{CH}_2-$), 1.95(3 H, d, $J=1.5$ Hz, $\text{CH}_3\text{C}=\text{C}$), and 5.5(1 H, q, $J=1.5$ Hz, $\text{CH}=\text{C}$). The yield was determined by GLC.
- 4) This compound was synthesized⁶⁾ by the reaction of isopropylidenecyclopropane and CO_2 in the presence of $\text{Pd}(\text{PPh}_3)_4$.
- 5) Mp 44.5-45.5 °C (lit⁷⁾ 44.5-46.5 °C). IR(KBr) 1750 and 1730(branched C=O), 1640(C=C), and 1280(C-O) cm^{-1} . NMR(CCl_4) δ 1.35(6 H, s, CH_3), 2.0(3 H, d, $J=1.4$ Hz, $\text{CH}_3\text{C}=\text{C}$), and 5.55(1 H, q, $J=1.4$ Hz, $\text{HC}=\text{C}$). $\text{UV}_{\text{max}}(\text{EtOH})$ 206 nm (ϵ 16,300). Found: C, 65.71; H, 7.93%. Calcd for $\text{C}_7\text{H}_{10}\text{O}_2$: C, 66.65; H, 7.99%.
- 6) Y. Inoue, T. Hibi, M. Satake, and H. Hashimoto, J. Chem. Soc. Chem. Commun., 1979, 982.
- 7) J. M. Stewart and D. W. Wooley, J. Am. Chem. Soc., 81, 4951(1959).

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